

**Supporting Information for:**

**The 6-31B Basis Set and the BMC-QCISD and BMC-CCSD  
Multi-Coefficient Correlation Methods**

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Table S1. Experimental Ionization Potentials and Electron Affinities (kcal/mol)

System	IP	EA	System	IP	EA
H		17.4 <sup>a</sup>	NH <sub>3</sub>	233.2	
Li	124.3	14.3	CN	313.7	89.0
Be	215.0		MgH		23.7
B	191.4	6.5	CF	209.4	
C	259.7	29.1	SiH	181.8	29.3
N	335.2		O <sub>2</sub>	278.9	10.8
O	313.9	33.7	HNO		6.8
F	401.8	78.4	PH	234.1	23.2
BeH		15.5	FO	294.7	51.9
Na	118.5	12.6	SH	238.9	53.3
Mg	176.3		PH <sub>2</sub>	226.3	29.4
Al	138.0	10.0	NaLi	116.8	
Si	187.9	31.9	NaF		11.9
P	241.9	17.2	LiCl		13.5
S	238.9	47.9	AlO	218.4	60.0
Cl	299.1	83.4	Na <sub>2</sub>	112.7	9.8
CH		28.2	PO	193.5	25.2
CH <sub>2</sub>		14.0	O <sub>3</sub>	288.9	47.5
NH		8.4	Al <sub>2</sub>	138.3	
CH <sub>3</sub>	226.0		MgCl	176.8	36.4
OH	299.1	42.1	S <sub>2</sub>	216.0	38.5
HF	369.7		Cl <sub>2</sub>	265.3	55.6
AlF	224.2		Al <sub>3</sub>		44.2

<sup>a</sup> This electron affinity is in Database/4MH but not in Database/4

Table S2. Best Estimate of Atomization Energies (kcal/mol)

Molecule	$D_e$	Molecule	$D_e$	Molecule	$D_e$
H <sub>2</sub>	109.48	H <sub>3</sub> CCH <sub>3</sub>	712.80	Si <sub>2</sub>	74.97
LiH	57.95	H <sub>3</sub> COH	512.78	NaCl	97.79
BeH	49.94	H <sub>2</sub> NNH <sub>2</sub>	438.60	MgS	54.43
Li <sub>2</sub>	24.36	HOOH	268.57	MgCl	74.46
CH(2P)	84.00	F <sub>2</sub>	38.20	P <sub>2</sub>	117.09
BeLi	6.69	CH <sub>3</sub> NH <sub>2</sub>	582.56	C <sub>4</sub> H <sub>6</sub> <sup>d</sup>	1012.37
CH <sub>2</sub> (3B <sub>1</sub> )	190.72	LiCl	113.90	C <sub>4</sub> H <sub>6</sub> <sup>e</sup>	1004.13
CH <sub>2</sub> (1A <sub>1</sub> )	181.37	NaF	114.00	C <sub>4</sub> H <sub>6</sub> <sup>f</sup>	987.20
NH	83.67	NaOH	187.77	C <sub>4</sub> H <sub>6</sub> <sup>g</sup>	1001.61
BH <sub>3</sub>	281.08	MgO	59.19	C <sub>2</sub> H <sub>2</sub> O <sub>2</sub> <sup>h</sup>	633.35
CH <sub>3</sub> (2A'' <sub>2</sub> )	307.46	PC	124.63	PO <sub>2</sub>	263.29
NH <sub>2</sub>	181.90	AlO	122.31	S <sub>2</sub>	101.67
OH	107.06	BeCl	90.20	SO <sub>2</sub>	257.86
CH <sub>4</sub>	420.11	MgF	110.86	BF <sub>3</sub>	470.04
NH <sub>3</sub>	297.90	MgOH	184.35	CH <sub>3</sub> NO <sub>2</sub>	601.27
H <sub>2</sub> O	232.60	CO <sub>2</sub>	389.14	HCOOCH <sub>3</sub>	785.26
HF	141.05	SiO	192.08	H <sub>3</sub> CCOOH	803.04
B <sub>2</sub>	67.14	SC	171.31	H <sub>3</sub> CCOCH <sub>3</sub>	977.96
LiO	82.00	C <sub>3</sub> H <sub>4</sub> <sup>a</sup>	704.79	C <sub>4</sub> H <sub>8</sub> <sup>i</sup>	1149.01
BC	100.00	C <sub>3</sub> H <sub>4</sub> <sup>b</sup>	703.20	C <sub>4</sub> H <sub>8</sub> <sup>j</sup>	1158.61
NaH	48.77	C <sub>3</sub> H <sub>4</sub> <sup>c</sup>	682.74	CINO	192.04
LiF	138.92	CH <sub>3</sub> CN	615.84	CH <sub>3</sub> CONH <sub>2</sub>	868.38
LiOH	211.68	H <sub>2</sub> CCO	532.32	(CH <sub>3</sub> ) <sub>3</sub> C	1199.34
BeO	104.77	N <sub>2</sub> O	270.22	Cl <sub>2</sub>	57.97
CN	180.58	HCP	250.50	Si <sub>2</sub> H <sub>6</sub>	530.81
CCH	267.83	Na <sub>2</sub>	17.03	CHF <sub>3</sub>	457.50
MgH	30.90	AlF	161.30	NF <sub>3</sub>	204.53

BeF	137.84	NaLiO	145.00	C <sub>2</sub> H <sub>5</sub> OCH <sub>3</sub>	1095.12
BeOH	216.86	BeF <sub>2</sub>	305.22	C <sub>4</sub> H <sub>10</sub> <sup>k</sup>	1303.04
BO	193.12	Be(OH) <sub>2</sub>	464.08	C <sub>4</sub> H <sub>10</sub> <sup>l</sup>	1301.32
HCCH	405.39	CH <sub>3</sub> CO	581.58	C <sub>4</sub> H <sub>4</sub> O	993.74
HCN	313.20	NO <sub>2</sub>	228.25	C <sub>4</sub> H <sub>5</sub> N	1071.57
CO	259.31	PO	141.55	C <sub>5</sub> H <sub>8</sub> <sup>m</sup>	1284.28
N <sub>2</sub>	228.46	SO(3)	125.00	AlF <sub>3</sub>	426.50
AlH	70.22	HCOOH	500.98	CINO <sub>2</sub>	262.60
BF	182.10	CH <sub>3</sub> CHO	677.03	C <sub>5</sub> H <sub>5</sub> N	1237.69
HCO	278.39	C <sub>2</sub> H <sub>4</sub> O	650.70	CF <sub>4</sub>	476.32
NO	155.22	C <sub>3</sub> H <sub>6</sub>	853.41	PF <sub>3</sub>	363.87
H <sub>2</sub> CCH	445.79	H <sub>3</sub> CCHCH <sub>2</sub>	860.61	C <sub>6</sub> H <sub>6</sub>	1367.56
SiH <sub>2</sub> ( <sup>1</sup> A <sub>1</sub> )	151.79	(C <sub>2</sub> H <sub>4</sub> )NH	718.81	C <sub>4</sub> H <sub>4</sub> S	962.73
SiH <sub>2</sub> ( <sup>3</sup> B <sub>1</sub> )	131.05	O <sub>3</sub>	146.10	ClF <sub>3</sub>	125.33
H <sub>2</sub> CCH <sub>2</sub>	563.47	NaCN	280.00	CF <sub>3</sub> CN	639.85
H <sub>2</sub> CO	373.73	ClO	64.49	C <sub>2</sub> F <sub>4</sub>	589.36
O <sub>2</sub>	119.99	C <sub>2</sub> H <sub>5</sub> O	698.64	BFCl <sub>2</sub>	371.21
SiH <sub>3</sub>	227.37	(CH <sub>3</sub> ) <sub>2</sub> CH	900.75	SiF <sub>4</sub>	574.35
PH <sub>2</sub>	153.20	ClF	61.36	BCl <sub>3</sub>	322.90
CH <sub>2</sub> OH	409.76	CH <sub>3</sub> Cl	394.64	CHCl <sub>3</sub>	343.18
SH	86.98	CH <sub>3</sub> SH	473.84	AlCl <sub>3</sub>	306.26
C <sub>2</sub> H <sub>5</sub>	603.75	HOCl	164.36	SF <sub>6</sub>	477.96
FO	53.40	H <sub>3</sub> COCH <sub>3</sub>	798.05	CCl <sub>4</sub>	312.74
SiH <sub>4</sub>	322.40	H <sub>3</sub> CCH <sub>2</sub> O		C <sub>2</sub> Cl <sub>4</sub>	466.28
PH <sub>3</sub>	242.55	H	810.36	SiCl <sub>4</sub>	384.94
H <sub>2</sub> S	182.74	C <sub>3</sub> H <sub>8</sub>	1006.87		
HCl	106.51	NCCN	500.89		
		Al <sub>2</sub>	31.10		

*a*propyne*b*Allene*c*cyclopropene*d**trans*-butadiene*e*2- butyne*f*bicyclobutane*g*cyclobutene*h*glyoxal*i*cyclobutane*j*isobutene*k*isobutane*l**n*-butane*m*spiropentane

Table S3. Barrier Heights (kcal/mol)

Reaction	$V_f^\ddagger$	$V_r^\ddagger$
1. $\text{Cl} + \text{H}_2 \rightarrow \text{HCl} + \text{H}$	8.7	5.6
2. $\text{OH} + \text{H}_2 \rightarrow \text{H} + \text{H}_2\text{O}$	5.7	22.0
3. $\text{CH}_3 + \text{H}_2 \rightarrow \text{H} + \text{CH}_4$	11.4	14.6
4. $\text{OH} + \text{CH}_4 \rightarrow \text{CH}_3 + \text{H}_2\text{O}$	6.7	20.2
6. $\text{H} + \text{H}_2 \rightarrow \text{H}_2 + \text{H}$	9.6	9.6
7. $\text{OH} + \text{NH}_3 \rightarrow \text{H}_2\text{O} + \text{NH}_2$	3.2	13.2
8. $\text{HCl} + \text{CH}_3 \rightarrow \text{Cl} + \text{CH}_4$	1.8	7.8
9. $\text{OH} + \text{C}_2\text{H}_6 \rightarrow \text{H}_2\text{O} + \text{C}_2\text{H}_5$	3.4	20.7
10. $\text{F} + \text{H}_2 \rightarrow \text{H} + \text{HF}$	1.8	33.2
11. $\text{OH} + \text{CH}_3 \rightarrow \text{O} + \text{CH}_4$	7.8	13.7
12. $\text{H} + \text{PH}_3 \rightarrow \text{PH}_2 + \text{H}_2$	3.2	25.5
13. $\text{H} + \text{ClH}' \rightarrow \text{HCl} + \text{H}'$	18.0	18.0
14. $\text{OH} + \text{H} \rightarrow \text{H}_2 + \text{O}$	10.1	13.1
15. $\text{H} + \textit{trans}\text{-N}_2\text{H}_2 \rightarrow \text{H}_2 + \text{N}_2\text{H}$	5.9	41.0
16. $\text{H} + \text{H}_2\text{S} \rightarrow \text{H}_2 + \text{HS}$	3.6	17.4
17. $\text{O} + \text{HCl} \rightarrow \text{OH} + \text{Cl}$	9.8	9.9
18. $\text{CH}_4 + \text{NH} \rightarrow \text{NH}_2 + \text{CH}_3$	8.4	22.7
19. $\text{C}_2\text{H}_6 + \text{NH} \rightarrow \text{NH}_2 + \text{C}_2\text{H}_5$	8.0	18.4
20. $\text{C}_2\text{H}_6 + \text{NH}_2 \rightarrow \text{C}_2\text{H}_5 + \text{NH}_3$	10.4	17.8
21. $\text{NH}_2 + \text{CH}_4 \rightarrow \text{CH}_3 + \text{NH}_3$	14.5	17.9
22. $s\text{-trans cis-C}_5\text{H}_8 \rightarrow s\text{-trans cis-C}_5\text{H}_8$	38.4	38.4